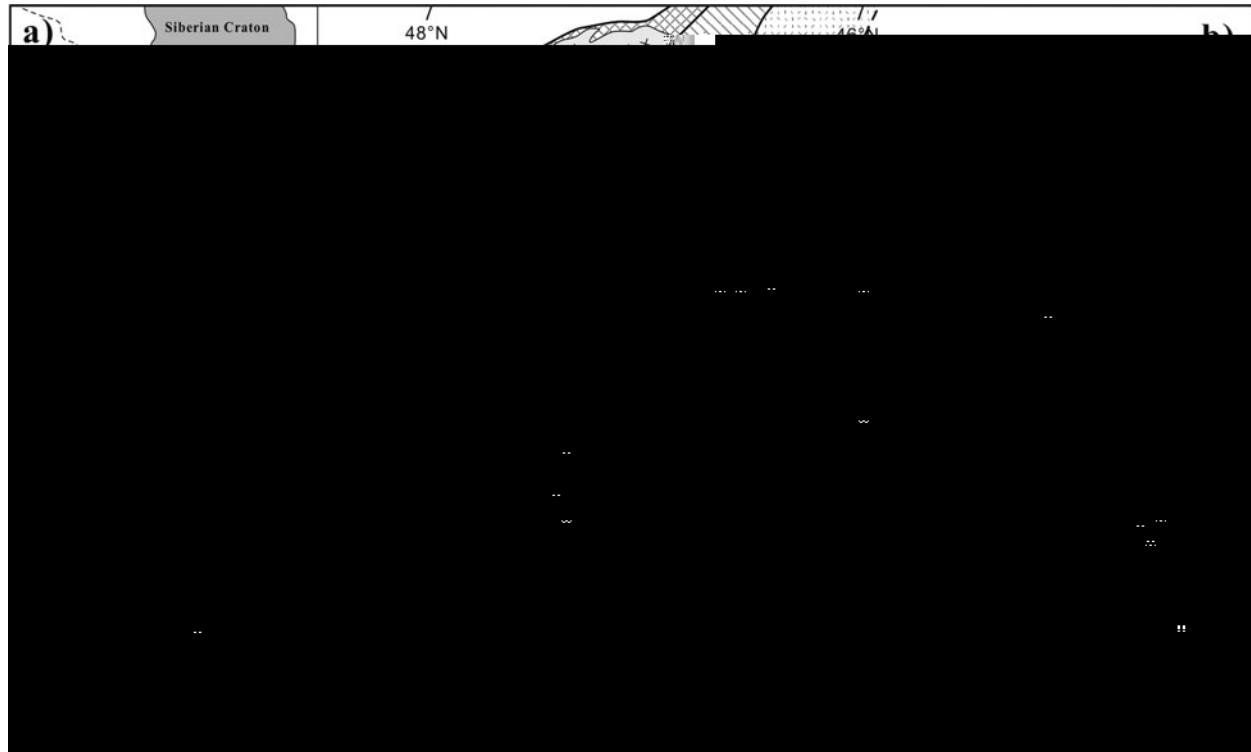


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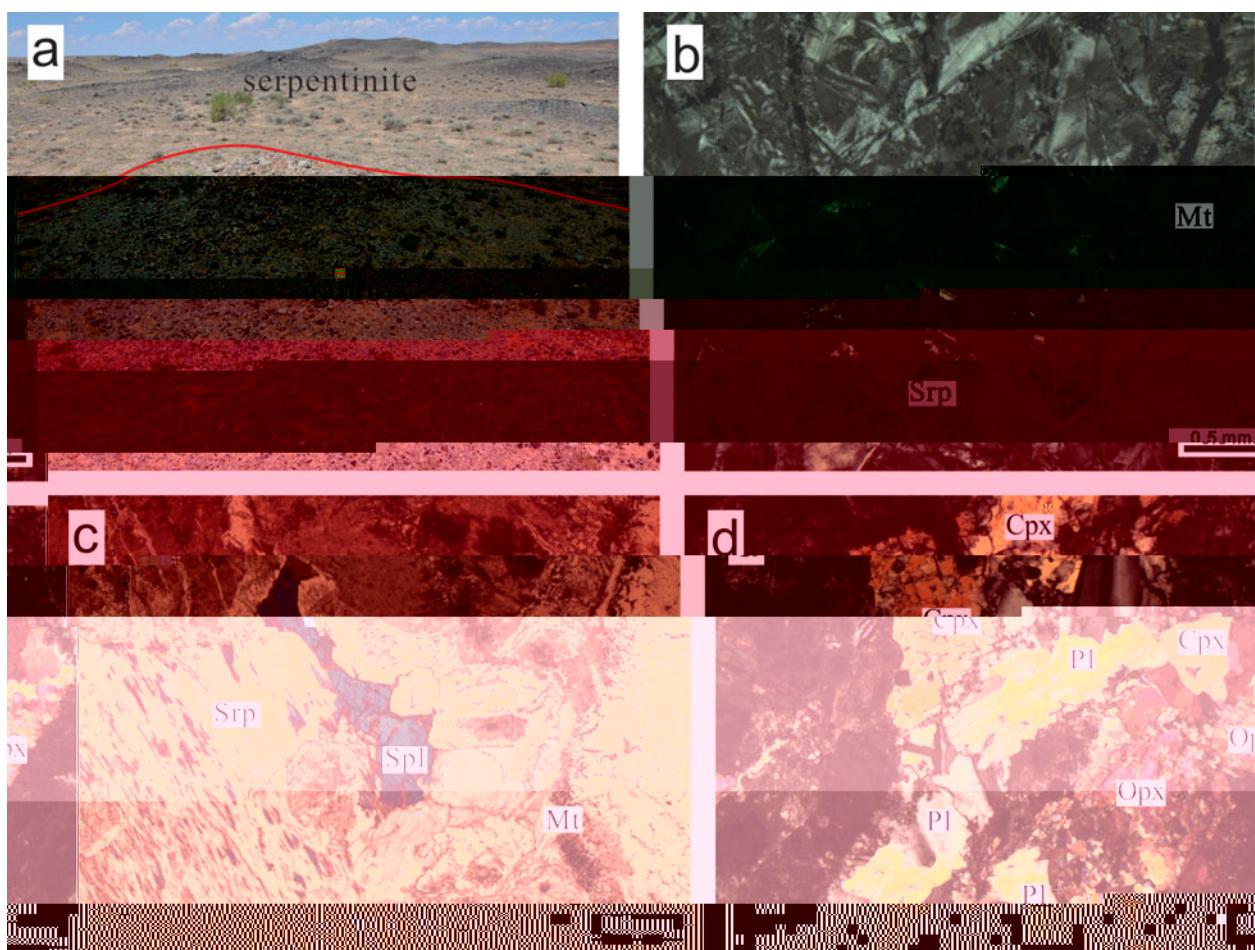
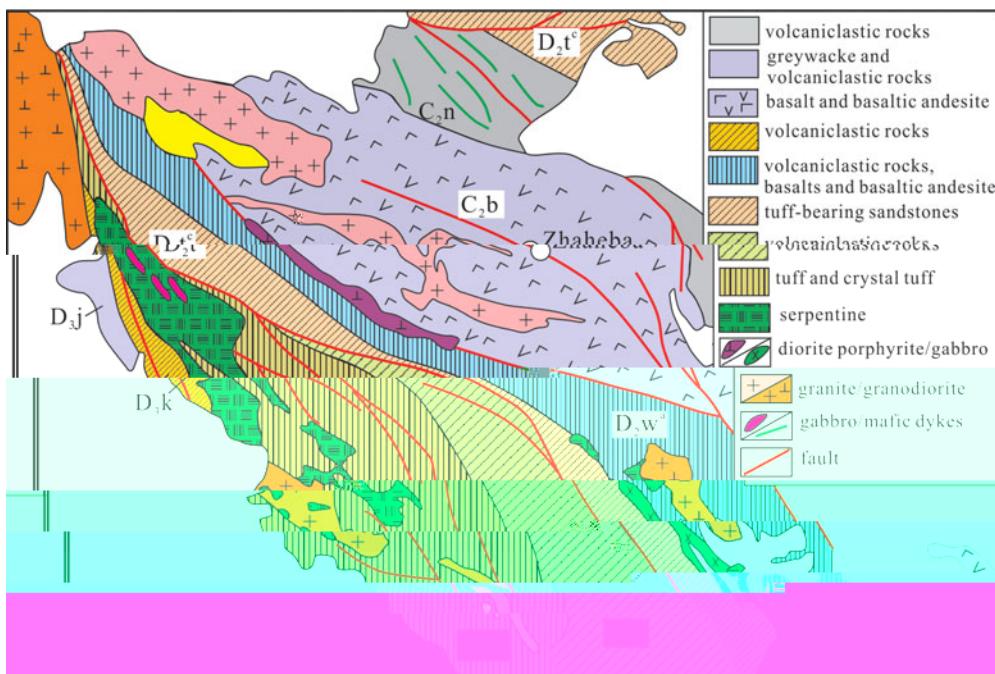
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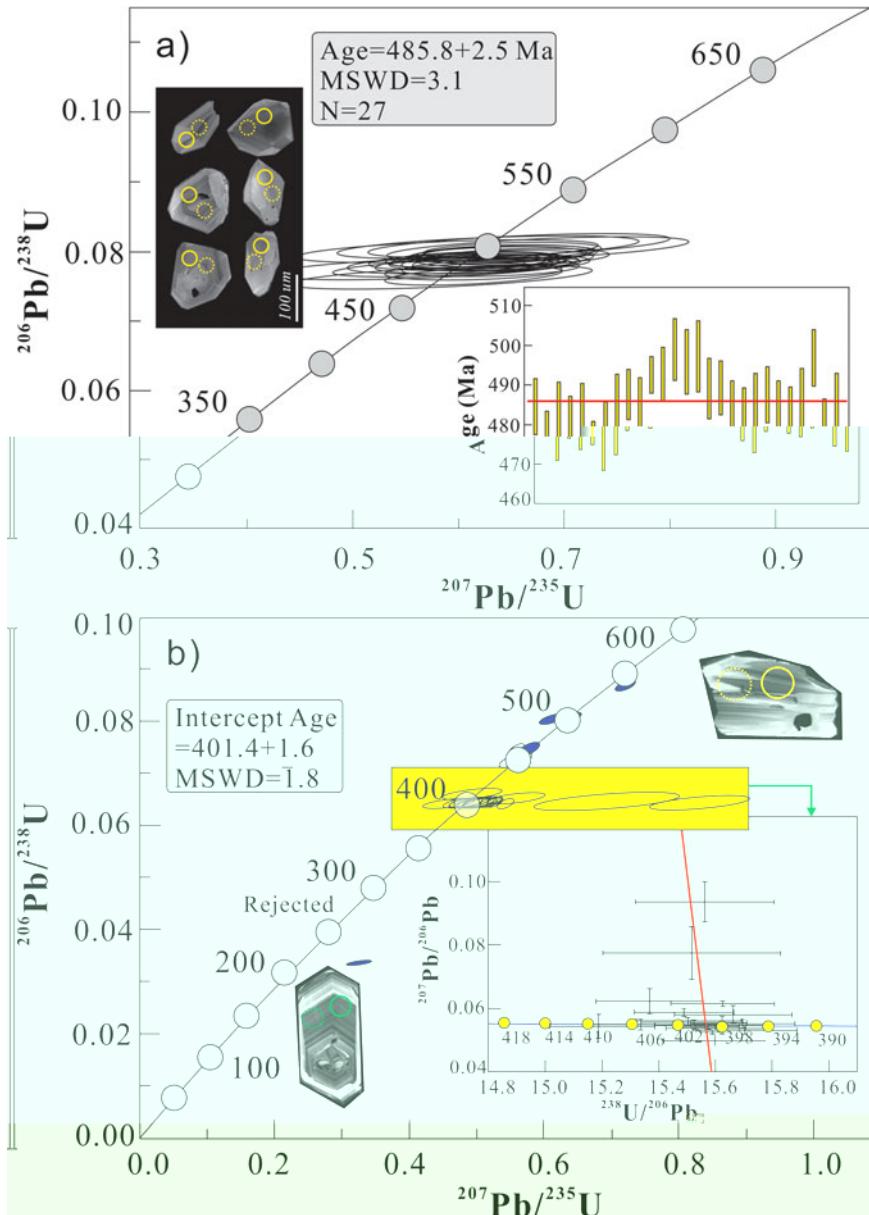
Table 1.

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et al. (2000a).

$$\varepsilon_{\text{ext}}(t) = 10000 \left(\frac{143}{144} \right)^t \left(\frac{143}{144} \right)^{(t-1)} \varepsilon_{\text{ext}}(t-1) \varepsilon_{\text{ext}}(t) \approx \left(\frac{143}{144} \right)^{401} \varepsilon_{\text{ext}}(0)$$



4. () 1% 2σ()

(± 4 , $= 2$, $= 3.1$). The error bars are $\pm 4\%$ ($\pm 1\%$ for the μ value) and the error bar for the μ value is $\pm 1\%$ (see *et al.* 2003).

4.b. M a c

4.b.1. Spinel composition

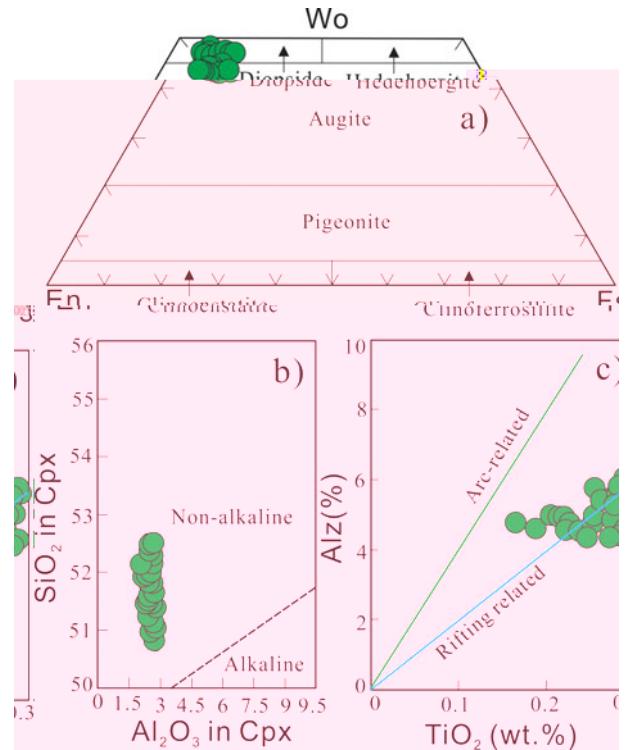
4.b.2. Pyroxene compositions

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4.c. W - c a c

4.c.1. Serpentinites and cumulates

(> 12%, 40%), (0.03–0.06%), (0.04–0.2%) (0.04–0.05%).



5. () 1% () 1% () 1% () 1% () 1% () 1% () 1% () 1%

45. % 51.2 %, 1 %
 $(3.24 \quad 4.6 \%)$, $(1.3 \quad 1.6\%)$, $(.54 \quad 15.42\%)$,
 2013 01-3), $(.12 \quad 0.34\%)$, $(2.1 \quad .3 \%)$, $(.11 \quad 0.46\%)$, $(.1 \quad 1\%)$
 2013 01-3) $(.11 \quad 0.46\%)$, $(.1 \quad 1\%)$

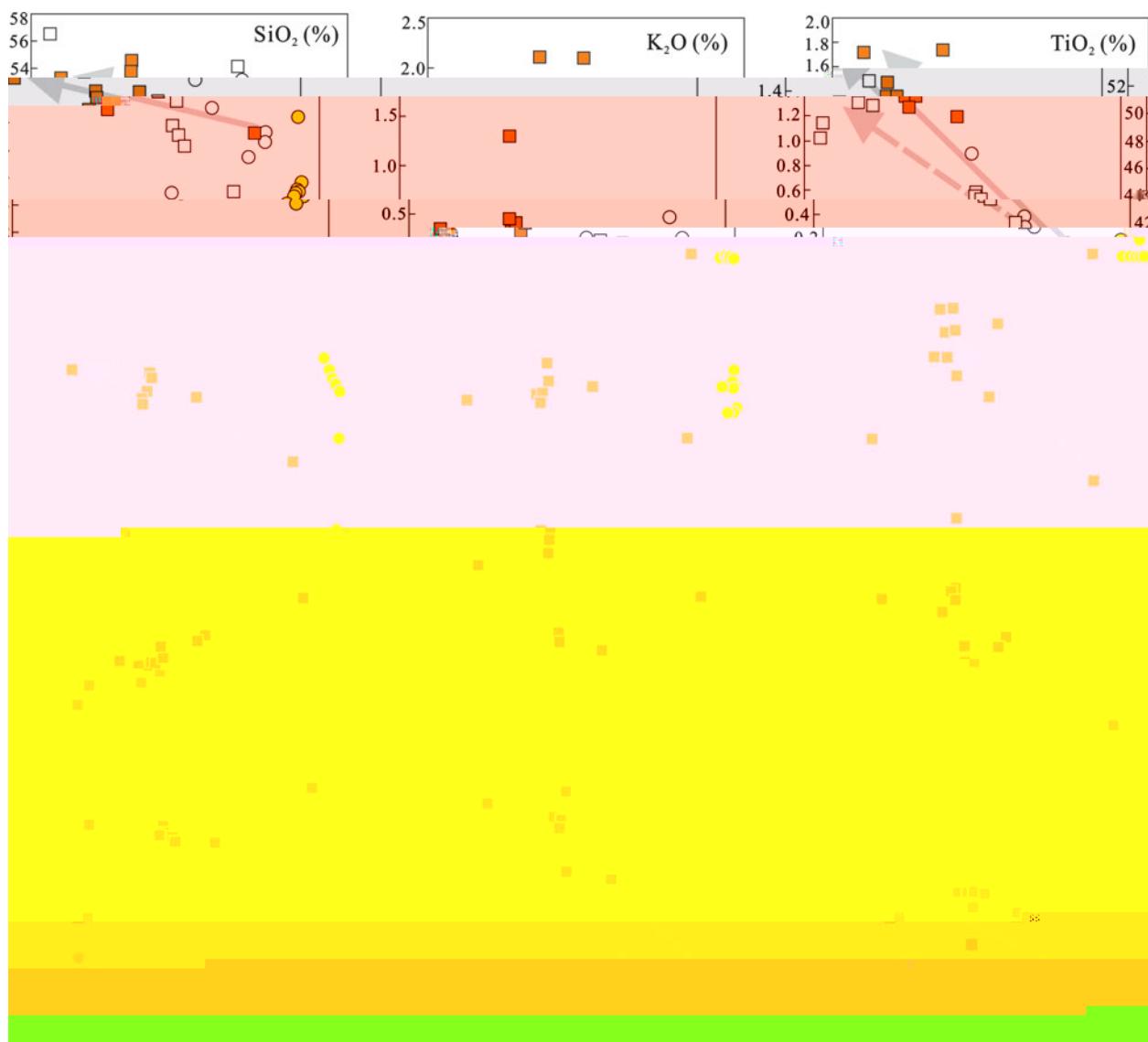
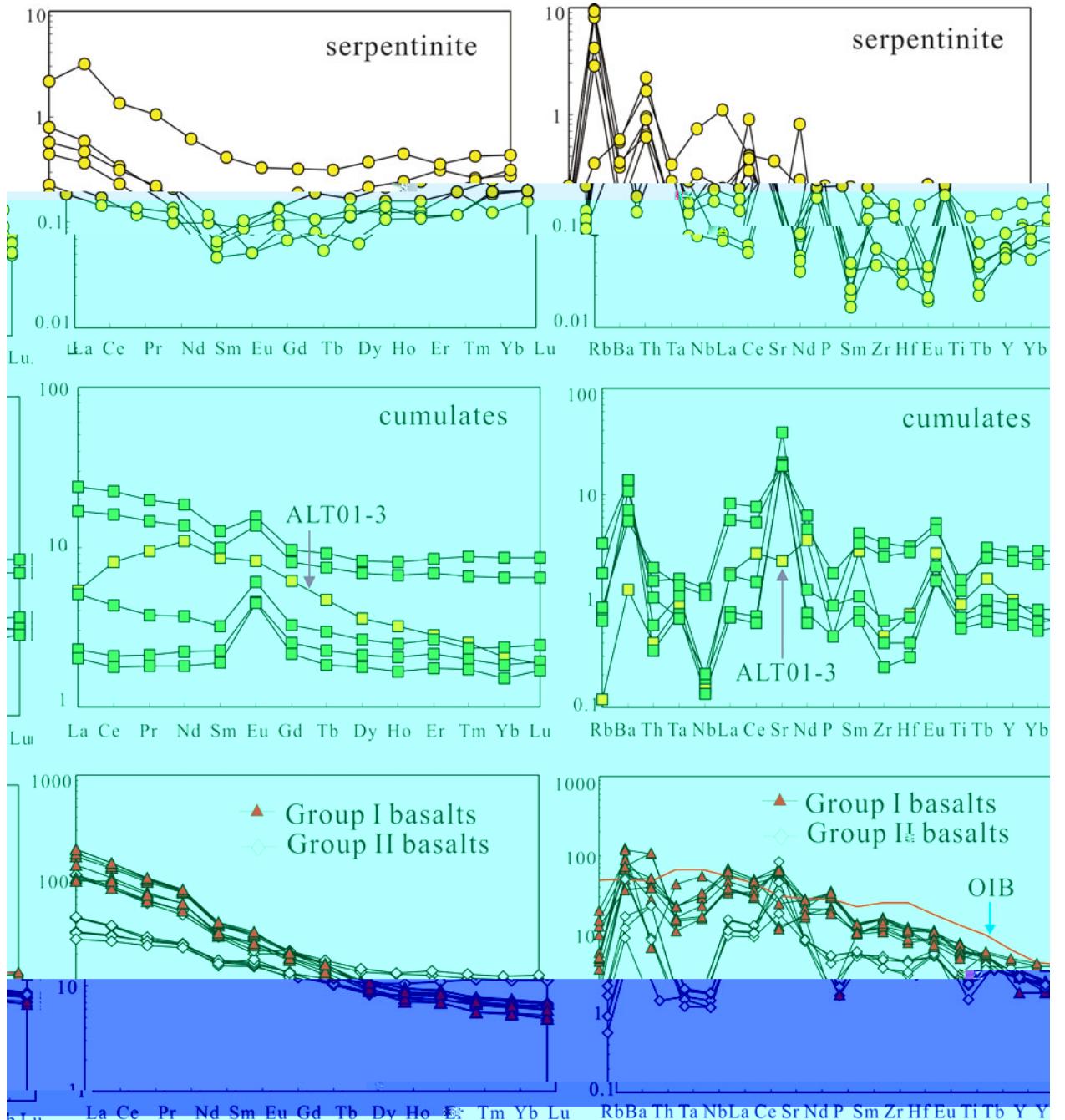


Fig. 6. (Continued) The effect of Fe^{2+} on the reduction of $\text{Cr}_2\text{O}_7^{2-}$ by Fe^{2+} at $1.0 \times 10^{-3} \text{ M}$ in 0.1 M H_2SO_4 at 25°C . The reaction was carried out in a thermostatic bath at 25°C for 1 h. The initial concentration of $\text{Cr}_2\text{O}_7^{2-}$ was $1.0 \times 10^{-3} \text{ M}$. The final concentration of $\text{Cr}_2\text{O}_7^{2-}$ was determined by the method of Fe^{2+} titration.

4.c.2. Basalts

43.15% 5.65% () % 52%,

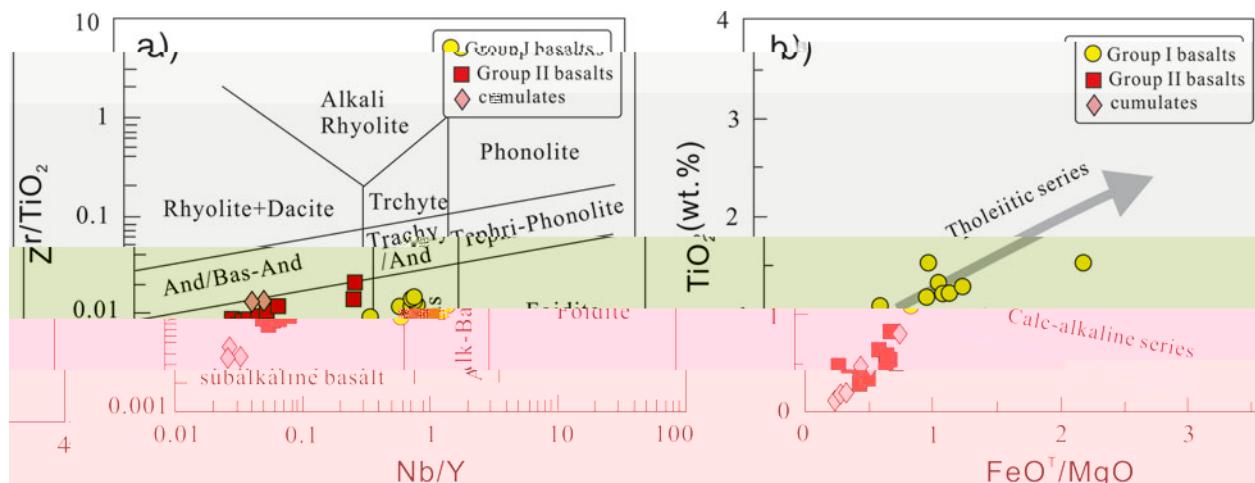
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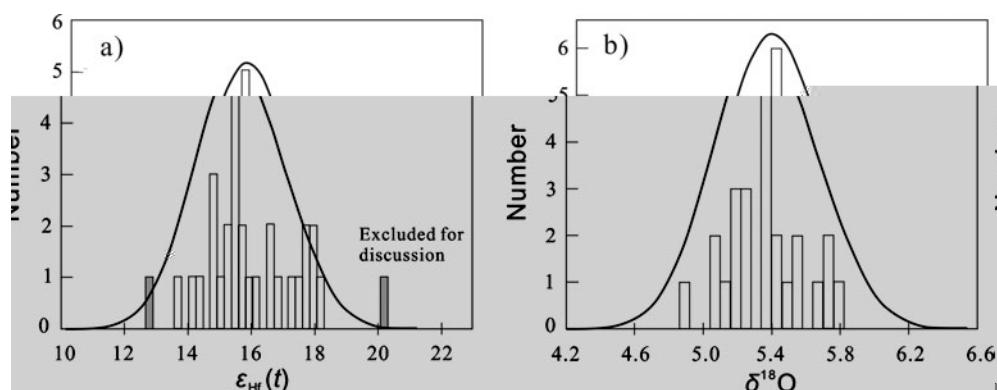
... & [\(L\)](#)).

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 0.0 , e_1 , e_2 , e_3 , e_4 , e_5 , e_6 , e_7 , e_8
 e_1 , e_2 , e_3 , e_4 , e_5 , e_6 , e_7 , e_8
 1.01 , e_1 , e_2 , e_3 , e_4 , e_5 , e_6 , e_7 , e_8
 e_1 , e_2 , e_3 , e_4 , e_5 , e_6 , e_7 , e_8 % 1 (~0.11). e_1 , e_2 , e_3 , e_4 , e_5 , e_6 , e_7 , e_8

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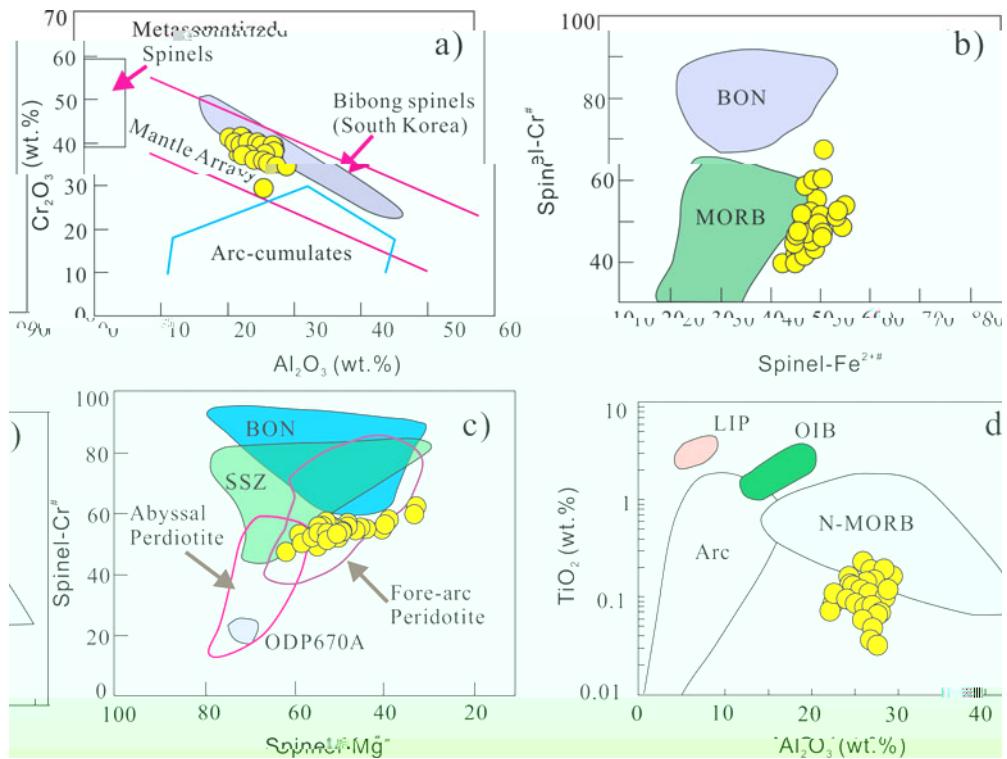
($\frac{1}{2} \times 100$)% = $\frac{1}{2}$ & $\frac{1}{2} \times 100$ = 50%.



$$e + (\dots) \varepsilon(t) + (\dots) = 1e - 1e + (\dots) e - e + 1\dots e.$$

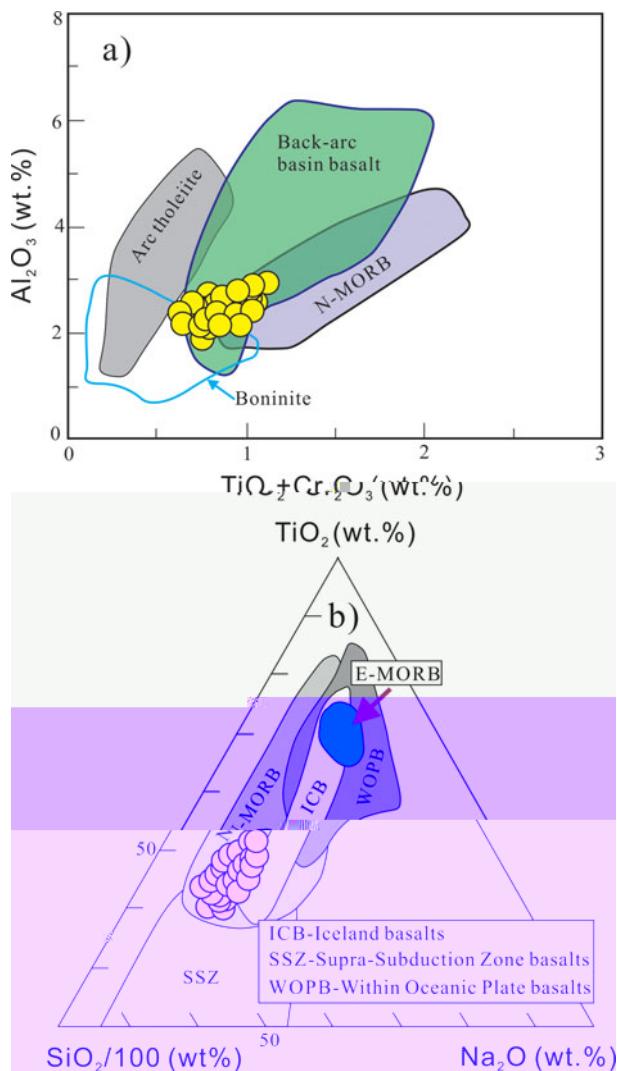
5.

5.a. T a Z a ba



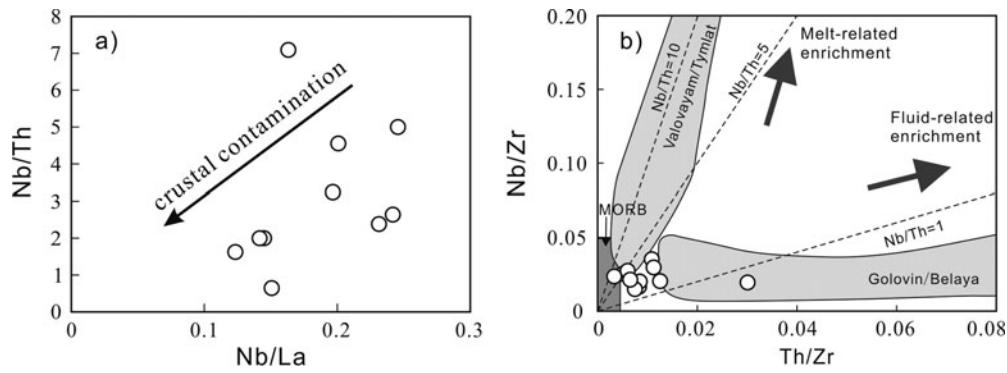
5.b. O

et al. 2010

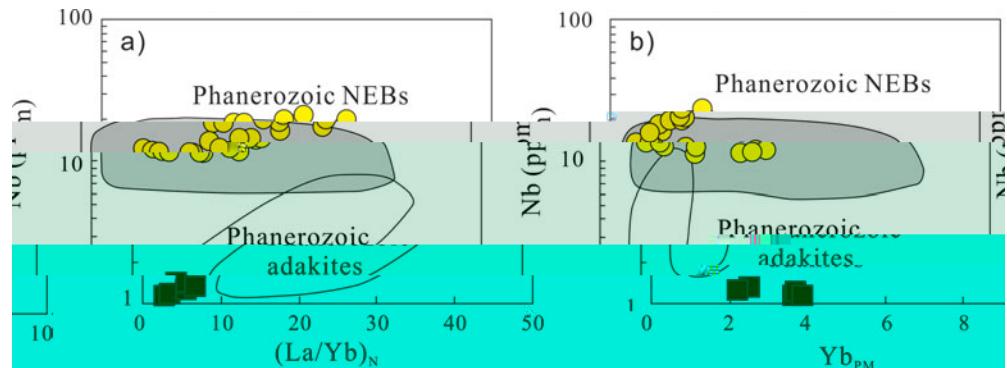


et al. (2002) (Fig. 12), which is 1% higher than the value obtained by the same authors in 1999 (Fig. 12). The difference between the two values is statistically significant at the 1% level.

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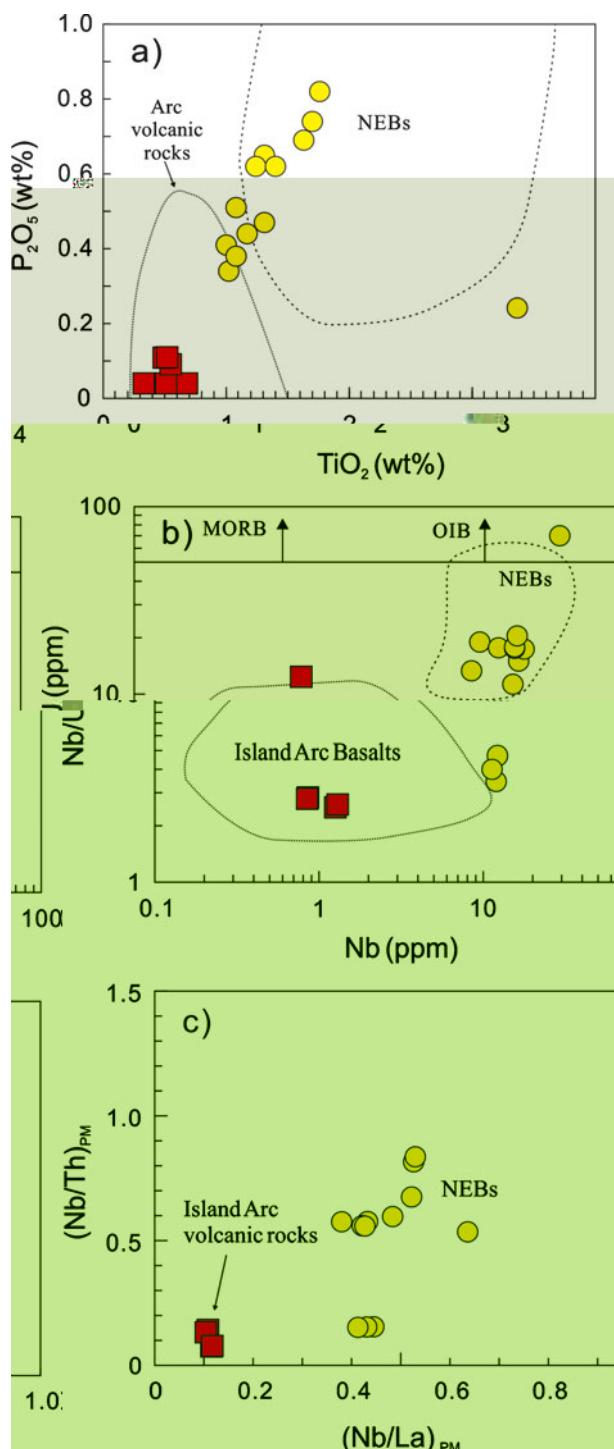
12. () *Levi-Civita symbol* ϵ_{ijk} ($i, j, k = 1, 2, 3$)



Exercice 13. (Continuation de l'exercice 12) Soit $\mathbf{e}_1 = \begin{pmatrix} 1 \\ 0 \\ 0 \end{pmatrix}$, $\mathbf{e}_2 = \begin{pmatrix} 0 \\ 1 \\ 0 \end{pmatrix}$, $\mathbf{e}_3 = \begin{pmatrix} 0 \\ 0 \\ 1 \end{pmatrix}$.

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et al. (195), 11% & (192)

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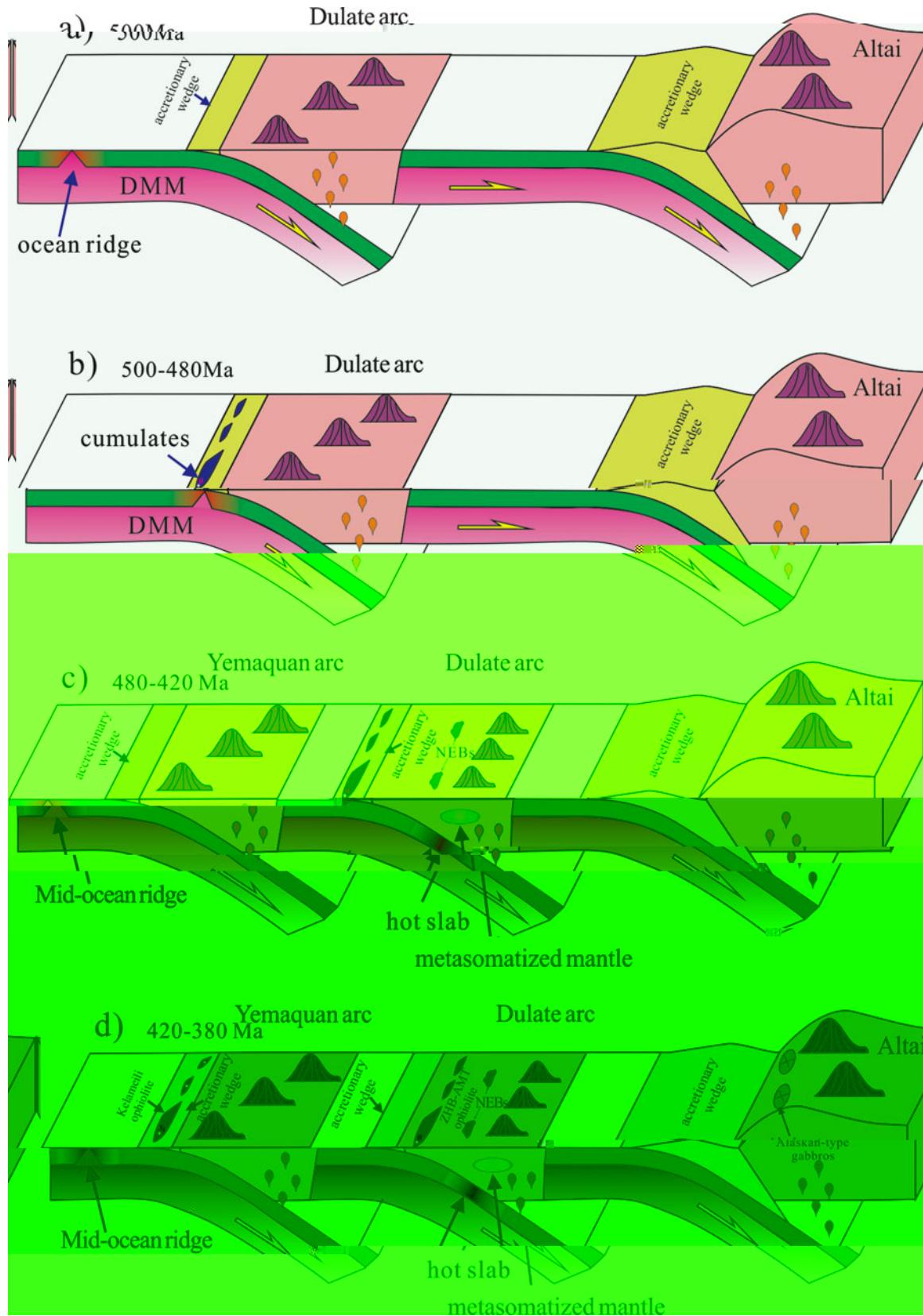


Figure 15. (a-d) Evolution of the Dulate arc; (e) evolution of the Yemaquan arc. (f-h) Evolution of the Dulate arc; (i-l) evolution of the Yemaquan arc. (m) Evolution of the Dulate arc; (n-p) evolution of the Yemaquan arc.

(4) *et al.* 2014) (*et al.* 2015). The 1% (420 3 0%). The 1% (400 3 0%). The 1% (420 3 0%). The 1% (400 3 0%). The 1% (420 3 0%). The 1% (400 3 0%). The 1% (420 3 0%). The 1% (400 3 0%).

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